

Synthesis of 8-Quinolinol Mannich Reaction Products

Sobia Mastoor¹, Kamran¹, Rubeena Saleem^{2,*}, Shaheen Faizi³

¹Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Hamdard University, Karachi, Pakistan ²Department of Chemistry, Faculty of Science, Jinnah University for Women, Karachi, Pakistan ³H.E.J. Research Institute of Chemistry, International Center for Chemical and Biological Sciences, University of Karachi, Karachi, Pakistan

Author's Contribution

All the authors contributed significantly to the research that resulted in the submitted manuscript.

Article info.

Received: May 19, 2018 Accepted: May 28, 2018

Funding Source: Nil Conflict of Interest: Nil

Cite this article: Mastoor S, Kamran, Saleem R, Faizi S. Synthesis of 8-Quinolinol Mannich Reaction. RADS J. Pharm. Pharm. Sci. 2018; 6(2): 138-143.

*Address of Correspondence Author: rs127pk@yahoo.com

ABSTRACT

Objective: To synthesize biologically significant molecules applying Mannich reaction conditions.

Method: Equimolar quantities of corresponding aldehydes and amines were mixed with 8-quinolinol (8-HQ) in absolute ethanol at room temperature and pressure without using any catalyst. The reaction progress was monitored through TLC.

Results: The condensation of privileged 8-quinolinol (8-HQ, **1**) with various substituted aldehydes and amines has resulted in the synthesis of ten products (**2-11**) under Mannich reaction conditions. These include four Mannich bases (**2-5**), three *bis*-products (**6-8**) and three imines (**9-11**). Nine of these derivatives are new and only one of them is known (**11**). The chemical characterizations of these molecules have been made through extensive study of ¹H, ¹³C NMR and 2D NMR techniques. The long range interaction of H-4 and H-5 with C-10 in HMBC plot, revised the assignment of C-9 and C-10 in 8-quinolinol moiety of newly synthesized derivatives.

Conclusion: Several new derivatives of 8-quinolinol have been prepared under Mannich reaction condition without any catalyst. Exact characterization of these new molecules will serve as a reference spectral data for future study. As Mannich products generally possess significant biological properties, future pharmacological study may turn them as an important target for drug synthesis.

Keywords: 8-Quinolinol, Retro-Mannich reaction, bis-product, Mannich base, imine.

INTRODUCTION

The ensemble of chemical functionality in Mannich bases plays a significant role as drug or drug predecessor [1,2]. Mannich bases have been reported as neuroprotective [3], anticancer. antimicrobial and antiparasitic agents [4]. Mannich base and its derivatives are frequently used in the production of natural products and macromolecules [2]. 8-Quinolinol (8-HQ, 1) is a well-known privileged heterocyclic scaffold. Many of 8-HQ derivatives have recognized been for its potency against neurodegenerative disorders as it is able to cross blood brain barrier and can selectively remove the Cu (II) and Zn (II) from amyloid beta-peptide and has capability to inhibit acetyl cholinesterase and monoamine oxidase [5]. 8-HQ polymeric micelles have promising activity against visceral leishmaniasis [6]. The 8-quinolinoyl jasmonate is found to inhibit different types of carcinomas [7] while gluconjugates have been reported to possess antiproliferative activity [8]. The invitro study of compounds having 8-HQ moiety with amino benzothiazole revealed IC 50 <1 M for the inhibition of DENV-2 protease in dengue [9]. virus type-2 Recently, diaza-18-crown-6hydroxyquinoline synthesized via Mannich reaction, is reported for sensitive Mg²⁺ analysis as a fluorescent probe for quantitative use within the cell [10] while the copper bis-products of 8-HQ found to possess anticancer activity [11]. Hence, the scope of 8-HQ (1) plus chemical functionality of Mannich reaction rationalize the use of 8-HQ Mannich products in future drug designing and drug discovery.

Keeping all this in view, different 8-HQ derivatives have been prepared applying Mannich reaction conditions. Current approach has resulted in the synthesis of Mannich bases (2-5), bis-products (6-8) and imines (9-10) which according to Science Finder research engine [12] have not been reported earlier.

METHODOLOGY

UV (in MeOH) and IR (in KBr disc) spectra were recorded on Thermo evolution 300 vision pro version 4.10 FTIR-8900 (Shimadzu, and Japan) respectively. spectrophotometers EI-MS was recorded on JMS 600H (Tokyo, Japan); HREI-MS on Thermo Finnigan MAT 95 XP (Germany); ESI-MS on Qstar XL/MS/MS system (USA). The ¹H and ¹³C NMR spectra were recorded in $CDCI_3$, $DMSO-d_6$ and acetone- d_6 at 24-27°C with Bruker Aspect AM-300, AM-400 and AM-500 spectrometers (Switzerland) working at 300, 400 and 500 MHz for ¹H NMR and 75, 100 and 125 MHz for ¹³C NMR, respectively. Exact assignment of ¹³C NMR spectral assignments was made partially through DEPT, HMQC and HMBC spectra and partially through literature values [13,14] and (ACD/Labs) software [15]. The precoated cards (0.2mm thickness) E. Merck Kiesel gel 60 GF₂₅₄ were used for thin layer chromatography. Chromatograms were visualized by UV at 254 and 365 nm. All the chemichydroals and solvents used were of analytical grade.

General Synthetic Procedures for Mannich Products

In each case, the measured quantity of corresponding aldehydes and amines (except for *bis*-products 6-8, where amine used was dimethyl amine for 6 & 8 and aniline for 7) were mixed with 8-HQ in equimolar proportion (0.005 M) in absolute ethanol at room temperature and pressure. The reaction mixture afforded products after 10 (9, 102mg), 13 (4, 638mg), 14 (10, 175mg; 11, 174mg), 30 (2, 136mg; 3, 500mg; 6, 64mg) and more than 60 (5, 866mg; 7, 386mg; 8, 889mg) days. The reaction progress was monitored through TLC and envisioned in UV light (366 nm and 254 nm). The crude products formed were separated from respective mother liquors, filtered and washed with methanol.

7-[α-(2"Methoxy, 5"-nitroanilino)-2'-thienyl]-8hydroxyquinoline (2)

UV λ_{max} MeOH nm: 248, 304. IR (KBr) v_{max} cm⁻¹: 3387 (OH), 1517 (C=C), 1337 (C-N), 1260 (C-O). HREIMS m/z: 407.0921 (M⁺, calculated for C₂₁H₁₇N₃O₄S, 407.18701). EIMS m/z (%): 407 (2), 238 (100), 168 (79), 122 (29). ¹H NMR: Table **1**, ¹³C NMR: Table **2**.

7-[α-(2"-Methoxy, 5"-nitro-anilino)-3'hydroxybenzyl]-8-hydroxyquinoline (3)

UV λ_{max} MeOH nm: 207, 247. IR (KBr) v_{max} cm⁻¹: 3634 (OH), 3544 (NH), 3086 (CH) 1580-1502 (C=C), 1380 (C-N), 1268 (C-O). EIMS m/z (%): 417 (1), 260 (10), 242 (15), 168 (100), 142 (19), 144 (9).¹H NMR: Table **1**, ¹³C NMR: Table **2**.

7-[α-(2", 5"-Dichloroanilino)-2', 3'dimethoxybenzyl]-8-hydroxyquinoline (4)

UV λ_{max} MeOH nm: 248, 307. IR (KBr) v_{max} cm⁻¹: 3620 (OH), 3363 (NH), 3004 (CH), 1589- 1508 (C=C), 1342 (C-N), 1188 (C-O), 729 (C-CI). HREIMS m/z: 454.0848 (M⁺, calculated for, C₂₄H₂₀Cl₂N₂O₃, 454.0853), 294.1106 (C₁₈H₁₆NO₃), 161.0453 (C₆H₄Cl₂N), 145.0542 (C₉H₇NO). EIMS m/z (%): 454 (8), 294 (36), 161 (100), 145 (5).¹H NMR: Table **1**, ¹³C NMR: Table **2**.

7-[α-(2",5"-Dichloroanilino)-3',4'dimethoxybenzyl]-8-hydroxyquinoline (5)

UV λ_{max} MeOH nm: 248, 307. IR (KBr) v_{max} cm⁻¹: 3620 (OH), 3360 (NH), 3005 (CH), 1570- 1510 (C=C), 1340 (C-N), 1180 (C-O). HREIMS m/z: 454.0847 (M⁺, calculated for, C₂₄H₂₀Cl₂N₂O₃, 454.0853). EIMS m/z (%): 454 (4), 294 (36), 161 (100), 145 (5).¹H NMR: Table **1**, ¹³C NMR: Table **2**.

7,7'-Benzyl-bis-8-hydroxyquinoline (6)

UV λ_{max} MeOH nm: 222, 250, 322. IR (KBr) v_{max} cm⁻¹: 3361 (OH), 3058 (NH), 3024 (CH), 1541-1508 (C=C), 1120 (C-O). HREIMS m/z: 378.1355 (M⁺, calculated for, C₂₅H₁₈N₂O₂, 378.1360). EIMS m/z (%): 378 (71), 301 (16), 145 (33). ¹H NMR: Table **1**, ¹³C NMR: Table **2**.

7,7'-4-lsopropylbenzyl-bis-8hydroxyquinoline (7)

UV λ_{max} MeOH nm: 222, 248. IR (KBr) v_{max} cm⁻¹: 3359 (OH), 3038 (NH), 3021 (CH), 2958 (aliphatic CH),

1600-1595 (C=C), 1120 (C-O). HREIMS m/z: 420.1800 (M⁺, calculated for, $C_{28}H_{24}N_2O_2$, 420.1839). EIMS m/z (%): 420 (32), 377 (2), 276 (33), 145 (14). ¹H NMR: Table **1**.

7,7'-Methylene-(3"-indolyl)-bis-8-hydroxyquinoline (8)

UV λ_{max} MeOH nm: 222, 248. IR (KBr) v_{max} cm⁻¹: 3392 (OH), 3012 (NH), 3010 (CH), 1649-1541 (C=C), 1342 (C-N). HREIMS m/z: 417.1477 (M⁺, calculated for, C₂₇H₁₉N₃O₂, 417.1478). EIMS m/z (%): 417 (100), 400 (4), 273 (20), 145 (16). ¹H NMR: Table **1**, ¹³C NMR: Table **2**.

2-methoxy-5-nitro-N-(benzylidine)-2'-hydroxy aniline (9)

UV λ_{max} MeOH nm: 229, 283. IR (KBr) v_{max} cm⁻¹: 3301 (OH), 3076 (CH), 1616 (C=N), 1580-1533 (C=C). HREIMS m/z: 272.0812 (M⁺, calculated for, C₁₄H₁₂N₂O₄, 272.0798). EIMS m/z (%): 272 (50), 241 (7), 179 (4), 165 (52), 120 (54), 77 (70). ¹H NMR: Table **1**, ¹³C NMR: Table **2**.

2-Methoxy-5-nitro-N-(benzylidine)-2', 5'-dihydroxy aniline (10)

UV λ_{max} MeOH nm: 244, 288, 373. IR (KBr) v_{max} cm⁻¹: 3749 (OH), 3055 (CH), 1615 (C=N), 1577-1510 (C=C), 1269 (C-O). EIMS m/z (%): 288 (100), 241 (22), 227 (18), 198 (11), 170 (13), 154 (96), 136 (40), 135 (41), 92 (19). ¹H NMR: Table **1**, ¹³C NMR: Table **2**.

2,5-Dihydroxy-N-(benzylidine) aniline (11)

UV λ_{max} MeOH nm: 247, 374. IR (KBr) v_{max} cm⁻¹: 3361 (OH), 3058 (CH), 1577 (C=N). HREIMS m/z: 213.0784 (M⁺, calculated for, C₁₃H₁₁NO₂, 213.0790). EIMS m/z (%): 213 (40), 104 (11), 93 (11). ¹H NMR: Table **1**.

RESULT AND DISCUSSION

Mannich bases (2-5) were produced by attack of 8quinolinol (1) on imine (formed as a result of reaction between aniline and aromatic aldehyde) in usual manner. However, some of the reaction combinations failed to generate Mannich bases, instead they gave 8-quinolinol *derivatives* can be related to the reaction between quinone methide (12a/12b/12c) and 1 formed from corresponding unstable Mannich bases

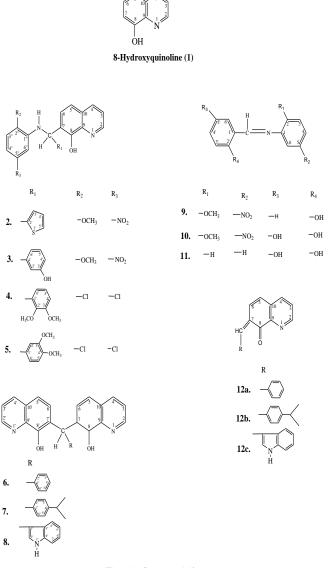
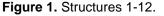


Figure 1. Structures 1-12



via deamination and retro Mannich reaction respectively [14,16].

NMR spectral data, particularly 2DNMR of these products revealed diagnostic and unambiguous assignments to the molecular structure. HMBC analysis of molecules altered the chemical shift values of C-9 and C-10 in 8-quinolinol moiety of Mannich bases (2-5) and bis derivatives (6 and 8). Current HMBC spectral study clearly indicated long range heteronuclear interaction of H-4 with C-10 at 138.09-138.27 in molecules 2-6 and 8 (Tables 1 and 2 are shown at the end of the article as an Annexure).

However, interaction between H-5 and C-10 was observed only in Mannich bases 4 and 5. Earlier, high frequency chemical shift was designated to C-10 and low frequency for C-9 in similar compounds [13,14].

It is noteworthy to mention that C 118.0 for C-3' in Mannich base (2) was not observed in 1D ¹³CNMR spectra (BB and DEPT). Its assignment is based on heteronuclear relationship between 7.39 (H-3') and 118.0 (C-3') in HSQC spectrum and long distant contact among 6.95 (H-4') and 118.0 (C-3') in HMBC plot. H-4' in (2) resonated as a multiplet at 6.95 and resolved into doublet of doublet ($J_{4', 3'}$ 4.3, $J_{4', 5'}$ 4.1 Hz) when shaked with D₂O. H-5' appeared as a broad singlet at 6.93 and converted into doublet ($J_{5',4'}$ 3.1

Hz) when treated with D₂O. 4"-Isopropyl substituent in *bis*-8-quinolinol (7) obviously illustrates the electron donating inductive effect and showed high field resonance at 7.08 (H-2", 6") and 7.02 (C-3", 5") in ¹HNMR spectrum as compared to corresponding protons at 8.19 (H-2", 6") and 7.24 (C-3", 5") in compound (6).

Among imines, 9 and 10 showed comparable chemical shifts for 2-methoxy-5-nitrophenyl imine moiety. As was not unexpected, H-4' appeared at high resonance (6.88) in 10 than corresponding proton in 9 due to presence of heteronuclear atom at adjacent carbon. ¹HNMR data of 11 presented in Table **1** showed clear multiplicity pattern as compared to the literature values [17].

Table 1. ¹H NMR chemical shifts and their multiplicities of compounds 2-11 in CDCI₃.

Assignment	2 ^a	3 ^a	4	5	6	7	8 ^b	9 ^a	10 ^a	11
2	0.074(0.0)	8.85dd	8.73dd	8.76dd	8.73dd	8.72m	8.77dd		_	7.34 d
	8.87d (3.0)	(4.5, 1.5)	(4.2,1.5)	(4.2, 1.5)	(4.2, 1.5)		(4.5, 1.5)	-	-	(8.0)
3	7.55m	7.54 m	7.38 dd	7.45 dd	7.37 dd	7.34 m	7.47 dd	7.34 d	7.32 d	7.48 dc
			(8.4, 4.2)	(8.4, 4.2)	(8.4, 4.2)		(8.5, 4.5)	(9.0)	(9.0)	(8.0,7.6
4	8.30 d (7.2)	8.28 dd	8.08 dd	8.11 dd	8.09 dd	8.09 m	8.27 dd	8.21 dd	8.19 dd	7.42 do
		(8.5, 1.5)	(8.4, 1.5)	(8.4, 1.5)	(8.4, 1.5)		(8.5, 1.5)	(9.0,2.5)	(9.0,3.0)	(8.0,7.6
5	7.43 m	7.54 m	7.27 d	7.31 d	7.30 d	7.35m	7.35 d	-	-	7.48 do
			(8.7)	(8.4)	(8.7)		(8.0)			(8.0,7.6
6	7.55 m	7.59 d	7.54 d	7.47 d	7.59 d	7.59 m	7.53 d	8.29 d	8.25 d	7.34 d
		(8.5)	(8.7)	(8.7)	(8.4)		(8.0)	(2.5)	(3.0)	(8.0)
2'	-	6.83 s	-	6.57 d	8.73dd	8.72m	8.77dd	_	-	
				(2.4)	(4.2, 1.5)		(4.5, 1.5)	_		
3'	7.39br.s	-	-	-	7.37 dd	7.34 m	7.47 dd	6.96 d	6.79 d	7.07 do
	7.0001.0				(8.4, 4.2)		(8.5, 4.5)	(8.5)	(8.5)	(8.5, 1.0
4'	6.95m	6.63 dd	6.84 dd (7.2, 2.4)	-	8.09 dd	8.09 m	8.27 dd	7.45dt	6.88dd	7.16 do
	0.3011	(8.0, 2.0)			(8.4, 1.5)		(8.5, 1.5)	(8.5,1.5)	(9.0, 3.0)	(8.5, 1.
5'	6.93br.s	7.12 t (8.0)	7.01 m	6.59 m	7.30 d	7.35m	7.35 d (8.0)	6.98ddd		
					(8.7)			(8.5,7.5,	-	-
					. ,			0.5)		
6′	-	6.87	7.01 m	6.92 dd	7.59 d	7.59 m	7.53 d	7.68dd	7.68 d	7.26 br
		d(7.5)	7.01.11	(8.1, 7.4)	(8.4)		(8.0)	(7.5,1.5)	(3.0)	
2″	-	-	-	-	8.19dd	7.08 m	7.31 br.s	-	-	-
-					(8.7,2.4)					
3″	7.03 d (8.4)	7.02 d	7.13 d	7.13 d	7.24 dd 7.02m	-	-	-	-	
5		(8.5)	(8.4)	(8.4)	(9.0,8.7)	1.02111				
4″	7.68 d (8.4)	7.38 d	6.54 dd	6.79 dd	7.05 dd	-	7.51		-	-
		(8.5)	(8.4, 2.4)	(8.4, 2.4)	(9.0,8.7)		d(6.5)			
5″	-	-	-	-	7.24 dd	7.02m	7.21 m	-	-	-
0					(9.0,8.7)					
6″	7.37 d (2.4)	7.22d (3.0)	6.60 d	6.99 d	8.19dd	7.08 m	7.25m	-	-	-
			(2.1)	(1.8)	(8.7,2.4)					
7"	-	-	-	-	-	-	7.41 d	-	-	-
							(6.0)			
СН	6.45d*	6.12 d*	6.43 d*	6.09 d*	6.72 s	6.89 s	7.01 s	9.08 s	8.96 s	8.76 s
	(7.8)	(6.4)	(6.6)	(5.4)						
NH	6.12d**(7.8)	5.84 d**	5.40 d** 5.19 d**		-	10.09	-	-	-	
	. ,	(6.4)	(6.6)	(5.4)			br.s**			
2"-OCH3	3.95s	3.95 s	3.78 s	3.83 [#] s	-	-	-	-	-	-
3"- OCH3	-	-	3.84 s	3.82 [#] s	-	-	-	-	-	
8-OH	10.33**s	9.35**s	NO	NO	NO	NO	10.03s**	-	-	-
2'-OH	-	-	-	-	-	-	-	13.18s**	12.34 s**	10.45s'
5'-OH	-	-	-	-	-	-	-	-	9.13 s**	8.01s*

^aSolvent (CD₃)₂SO, ^bSolvent(CD₃)₂CO, *Changed to singlet when shaked with D₂O; ** Disappeared when shaked with D₂O; # Exchangeable; NO: Not observed; **7**CH (CH₃)₂, \Box 2.86m;CH (CH₃)₂, \Box 1.19 d (6.9); **9** 2-OCH₃ \Box 3.99s; **10** 2-OCH₃ \Box 3.94s.

Assignments	2 ^a	3 ^a	4	5	6	8 ^b	9 ^a	10 ^a
1	-	-	-	-	-	-	137.23	137.74
2	148.45	148.36	148.14	148.22	147.92	148.42	157.94	157.96
3	122.06	121.86	121.77	121.94	121.41	122.68	112.19	112.13
4	136.15	136.03	136.07	136.05	135.88	136.93	123.64	123.51
5	113.73	113.21	117.78	118.19	117.25	118.04	140.94	140.95
6	126.32	126.42	126.93	126.89	128.48	122.99	114.53	114.53
7	123.36	123.77	122.95	122.88	124.75	123.49	-	-
8	152.12	151.89	152.92	154.52	151.10	149.62	-	-
9	128.06	127.69	127.80	127.74	129.11	127.05	-	-
10	138.20	138.09	138.23	138.21	138.18	139.78	-	-
1′	-	137.22	134.73	133.90	-	-	119.22	119.22
2′	117.74	113.82	147.00	117.41	147.92	148.42	160.56	153.43
3′	*118.00	157.45	149. 40	148.48	121.41	122.68	116.70	117.32
4'	126.92	114.16	112.16	149.06	135.88	136.93	133.78	121.79
5′	125.31	129.54	120.02	112.40	117.25	118.04	119.10	149.57
6′	-	117.62	124.02	119.13	128.48	122.99	132.76	117.32
7'	-	-	-	-	124.75	123.49	-	-
8′	-	-	-	-	151.10	149.62	-	-
9'	-	-	-	-	129.11	127.05	-	-
10′	-	-	-	-	138.18	139.78	-	-
1″	141.37	141.37	144.00	143.99	142.81	-	-	-
2″	136.85	149.98	117.45	116.93	128.31	129.67	-	-
3″	109.27	109.01	129.65	129.58	129.54	117.56	-	-
4″	126.92	117.70	117.13	111.23	126.56	122.21	-	-
5″	145.92	143.14	133.59	129.57	129.54	119.99	-	-
6″	103.90	103.48	112.10	110.59	128.31	122.52	-	-
7″	-	-	-	-	-	112.94	-	-
8″	-	-	-	-	-	138.27	-	-
9″	-	-	-	-	-	137.89	-	-
СН	51.30	55.05	51.94	55.93	45.02	37.21	165.18	164.88

δ 60.73, 3'-OCH₃ δ 55.79; **5,** 3'-OCH₃ δ 55.93, 4'-OCH₃ δ 55.88; **9,** 2-OCH₃δ 56.85; **10,** 2-OCH₃ δ 56.83.

CONCLUSION

Several new compounds comprising amine and 8-Quinolinol (8-HQ) have been generated under Mannich reaction condition without any catalyst. Revised spectral values and new spectral data may serve as a reference for similar compounds. Further study on pharmacology of these small molecules may refurbish them as a key for modern drug discovery.

REFERENCES

- Roman G. Mannich bases in medicinal chemistry and drug design. Eur J Med Chem. 2015;89:743-816.
- 2. Heravi MM, Lashaki TB, Fattahi B, Zadsirjan V. Application of asymmetric Sharpless aminohydroxylation in total synthesis of natural

products and some synthetic complex bio-active molecules. RSC Adv. 2018;8(12):6634-59.

- Li Y, Qiang X, Luo L, Yang X, Xiao G, Liu Q, et al. Aurone Mannich base derivatives as promising multifunctional agents with acetylcholinesterase inhibition, anti-β-amyloid aggragation and neuroprotective properties for the treatment of Alzheimer's disease. Eur J Med Chem. 2017;126:762-75.
- Mahal K, Ahmad A, Schmitt F, Lockhauserbäumer J, Starz K, Pradhan R, et al. Improved anticancer and antiparasitic activity of new lawsone Mannich bases. Eur J Med Chem. 2017; 126:421-31.
- Xu H, Chen W, Zhan P, Liu X. 8-Hydroxyquinoline: a privileged structure with a broad-ranging pharmacological potential. Med Chem Comm. 2015;6(1):61-74.
- Duarte MC, Lage LM, Lage DP, Martins VT, Carvalho AM, Roatt BM, et al. Treatment of murine visceral leishmaniasis using an 8-

hydroxyquinoline-containing polymeric micelle system. Parasitol Int. 2016; 65(6): 728-36.

- Ksshman Y, Flescher E, Herzberg M. Chemical derivatives of jasmonate, pharmaceutical compositions and methods of use thereof. PCT Int Appl Publ. 2013; US 20130203689 A1 20130808.
- Oliveri V, Giuffrida ML, Vecchio G, Aiello C, Viale M. Gluconjugates of 8-hydroxyquinolines as potential anti-cancer prodrugs. Dalton Trans. 2012;41(15):4530-5.
- Lai H, Prasad GS, Padmanabhan R. Characterization of 8-hydroxyquinoline derivatives containing aminobenzothiazole as inhibitors of dengue virus type 2 protease in vitro. Antiviral Res. 2013;97(1):74-80.
- Sargenti A, Farruggia G, Zaccheroni N, Marraccini C, Massimo S, Cappadone C, et al. Synthesis of a highly Mg2+-selective fluorescent probe and its application to quantifying and imaging total intracellular magnesium. Nat Protoc. 2017;12(3):461-71.
- 11. Weerasuriya DR, Wijesinghe WP, Rajapakse RM. Encapsulation of anticancer drug copper bis(8hydroxyquinoline) in hydroxyapatite for pH-

sensitive targeted delivery and slow release. Mater Sci Eng C. 2017;71:206-13.

- 12. http://www.sci.finder.cas.org.com.
- Mastoor S, Faizi S, Saleem R, Siddiqui BS. NMR study of O and N, O-substituted 8-quinolinol derivatives. Magn Reson Chem. 2014;52(3):115-21.
- 14. Faizi S, Saleem R, Siddiqui BS, Gilani AH, Zia-ul-Haq A, Saeed SA, Khan KA, Ahmad A Proc Pak Acad Sci. 2013;30: 231-46.
- Predicted NMR data calculated using Advanced Chemistry Development, Inc. (ACD/ Labs) software V4.04 (© 1994-1999 ACD/Labs).
- M. Tramontini, L. Angiolini. Further advances in the chemistry of mannich bases. Tetrahedron. 1998; 46:1791-1837.
- 17. Issam AM, Hena S. New T-shape polyurethanes: synthesis, thermal and mechanical properties. Polym Plast Technol Eng. 2012;51(4):391-5.



This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.